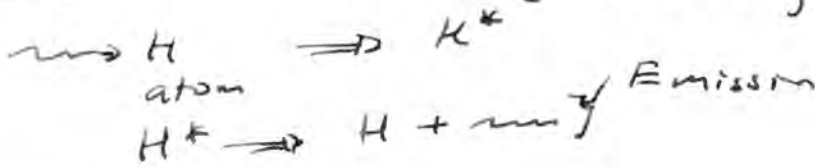
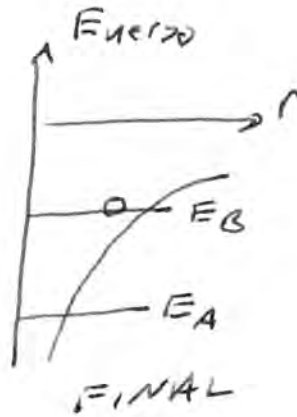
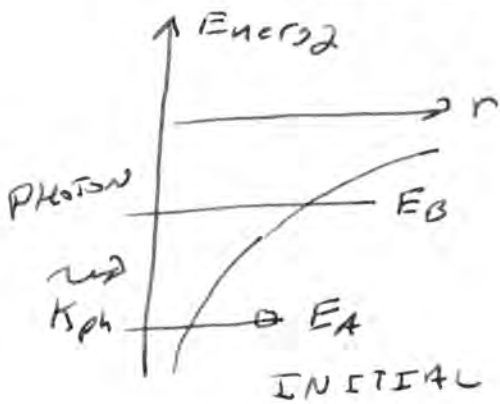




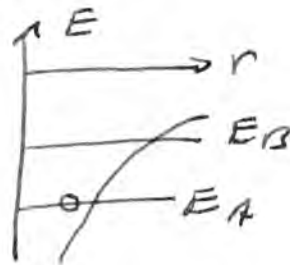
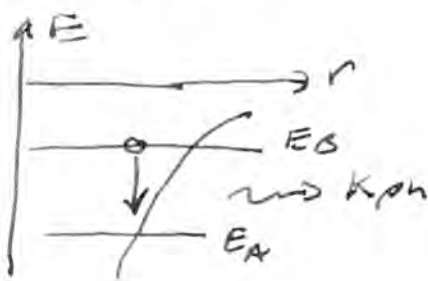
Light can be ~~absorbed~~ or emitted by matter
Example absorbed \leftarrow excited } absorption



ENERGY Levels



$K_{\text{photon}} = E_A - E_B$
 ONLY PHOTONS WITH THIS PRECISE ENERGY WILL BE ABSORBED



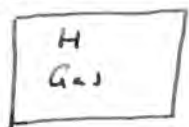
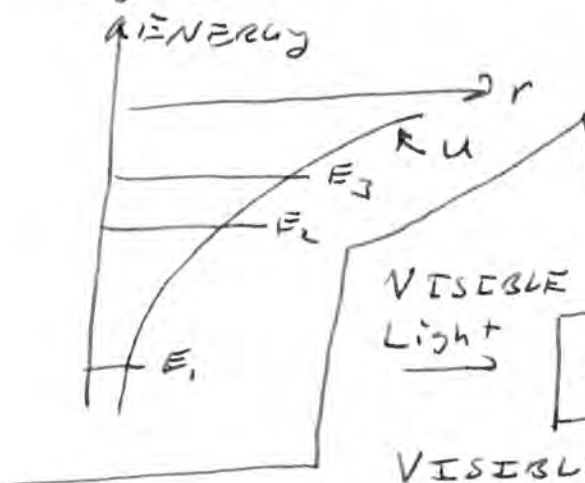
PHOTONS NOT WITH THIS ENERGY CANNOT BE EMITTED OR ABSORBED

HYDROGEN ATOM

$$E_N = -\frac{13.6}{N^2} \text{ (eV)}$$

$$U = -\frac{e^2}{4\pi\epsilon_0 r}$$

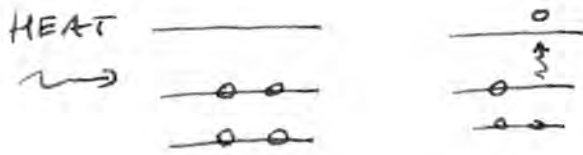
Other atoms also have energy levels, but with different spacings.



VISIBLE LIGHT 1.5-3 eV cannot be absorbed or emitted

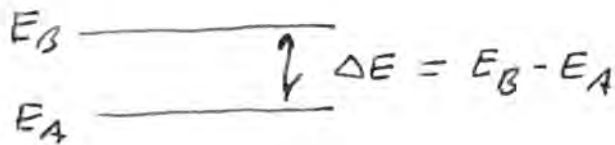
ASIDE: TEMPERATURE \Rightarrow CAN ALSO EX

L20
pl.5



FOR "NORMAL" TEMPERATURES THIS IS NOT A LARGE EFFECT

STATISTICAL DISTRIBUTION \Rightarrow MANY atoms
($\approx 10^{23}$ is "MANY, MANY..." ATOMS!)



$N(E_A)$ = NUMBER OF ATOMS WITH ENERGY E_A

$N(E_B)$ = NUMBER WITH E_B

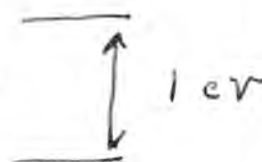
$$\frac{N(E_B)}{N(E_A)} = \exp(-\Delta E / k_B T)$$

$$k_B = 1.38 \times 10^{-23} \text{ J/K}$$

T = Temp. in K

EXAMPLE

$\Delta E = 1 \text{ eV}$



$T = 300 \text{ K}$

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$$

$$\frac{N(E_B)}{N(E_A)} = \exp\left(-\frac{1.6 \times 10^{-19}}{(1.4 \times 10^{23})(300)}\right)$$

$$= \exp(-38) = 3 \times 10^{-17}$$

Suppose

$$N(E_A) = 1 \text{ mole}$$

$$N(B) \approx 2 \times 10^7 \text{ atoms}$$

[REALLY SMALL NUMBER RELATIVE TO 10^{23} , BUT NOT ZERO]

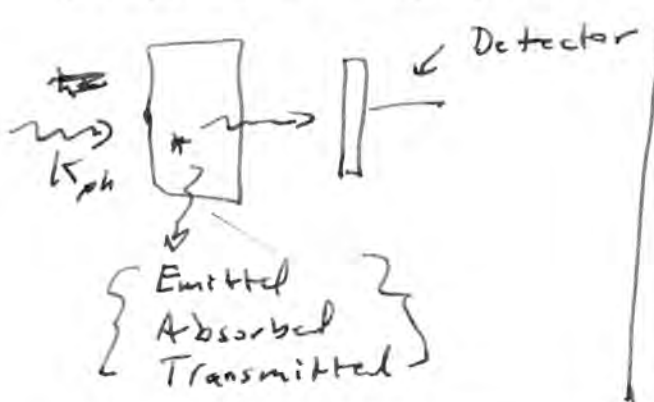
Suppose

$$T = \underline{5000 \text{ K}}$$

$$N(E_B) = \underline{0.1 \text{ mole}}$$

$$N(E_A) = 1 \text{ mole}$$

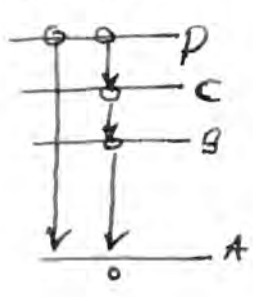
MEASURING ABSORPTION



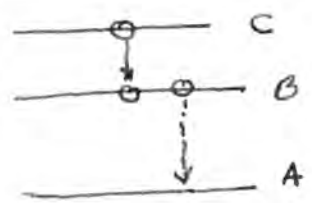
LIGHT ABSORBED
CAN HEAT UP
THE SOLID

L20
P.2

MULTIPLE EMISSION



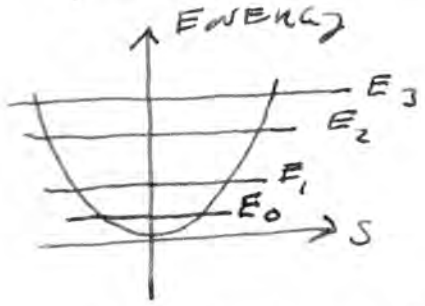
PHOTONS
 E_{AB}, E_{BC}, E_{CD}
CAN BE
OBSERVED.
~~AND~~ ALSO
 E_{AD}



SELECTION RULES:
SOME TRANSITIONS
ARE MORE LIKELY
TO OCCUR

THIS IS A GEOMETRY
ISSUE (PHYSICS IS MOSTLY
SYMMETRY AND ENERGY!)

VIBRATION ENERGY LEVELS



$$E = \frac{1}{2}mv^2 + \frac{1}{2}k_s s^2$$

DEFINE $\omega_0 = \sqrt{\frac{k_s}{m}}$

$$E_N = N \hbar \omega_0 + \frac{1}{2} \hbar \omega_0$$

$$= (N + \frac{1}{2}) \hbar \omega_0, \quad N = 0, 1, 2, 3, \dots$$

ENERGY LEVELS
UNIFORMLY
SPACED

\hbar = Planck's constant
(h) ~~divided~~ divided
by 2π . [called
"h-bar"]

$$\hbar = \frac{h}{2\pi} = 1.05 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$h = 6.6 \times 10^{-34} \text{ J}\cdot\text{s}$$

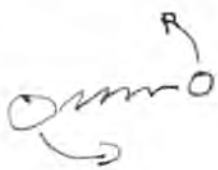
\hbar is VERY SMALL,
BUT IF $\hbar = 0$
NO QUANTUM
MECHANICS.

WEIRD RESULT: EVEN
AT TEMPERATURE = ZERO,
WITH VIBRATIONAL ENERGY
LEVELS ARE
STILL MOVING! $E_N > 0$!

ROTATIONAL ENERGY

L20

P.3



MOLECULE CAN ROTATE

LEVELS CLOSELY SPACED AND INCREASE AS ENERGY INCREASES



Electronic $\sim 1\text{eV} \sim 10^0 - 10^1 \text{ eV}$
 Vibrational $\sim 0.1\text{eV} \sim 10^{-1} \text{ eV}$
 Rotational $\sim 0.0001 \text{ eV} \sim 10^{-4} \text{ eV}$

START CPT. 9

END CHAPTER 8

READ 8.6
8.7

MULTIPARTICLE SYSTEMS

RECALL

$$\frac{d\vec{P}_{SYS}}{dt} = \vec{F}_{NET}$$

FOR VLLC

$$\vec{P}_{SYS} = M_T \vec{v}_{CM}$$

REVIEW

$$\vec{r}_{CM} = \frac{m_1 \vec{r}_1 + m_2 \vec{r}_2 + \dots}{M_T} \quad ; \quad M_T = m_1 + m_2 + \dots$$

REARRANGE AND TAKE DERIVATIVE

$$M_T \vec{r}_{CM} = m_1 \vec{r}_1 + m_2 \vec{r}_2 + m_3 \vec{r}_3 + \dots$$

$$M_T \vec{v}_{CM} = m_1 \vec{v}_1 + m_2 \vec{v}_2 + m_3 \vec{v}_3 + \dots$$

\vec{P}_{SYS}

CAN DO THIS SUM IN "BUNCHES" OF ATOMS



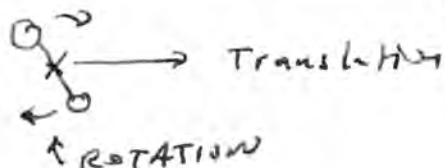
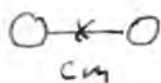
$$\vec{r}_{CM} = \frac{(m_{11} \vec{r}_{11} + m_{12} \vec{r}_{12} + \dots) + (m_{21} \vec{r}_{21} + m_{22} \vec{r}_{22} + \dots) + (m_{31} \vec{r}_{31} + m_{32} \vec{r}_{32} + \dots)}{(m_{11} + m_{12} + \dots) + (m_{21} + m_{22} + \dots) + (m_{31} + m_{32} + \dots)}$$

$$\vec{r}_{CM} = \frac{M_1 \vec{r}_1 + M_2 \vec{r}_2 + M_3 \vec{r}_3 + \dots}{M_1 + M_2 + M_3 + \dots}$$

KINETIC ENERGY

= ROTATIONAL + TRANSLATIONAL

+ VIBRATIONAL ENERGY



[FOR MOST OBJECTS \rightarrow BASKETBALL \Rightarrow FAST BALL + CURVE BALL]

$$K_{\text{Trans}} = \frac{1}{2} M_T v_{\text{cm}}^2 = \frac{p_{\text{cm}}^2}{2M_T}$$

POTENTIAL ENERGY

$$U_g = M_T g y_{\text{cm}}$$

WE HAVE BEEN DOING THIS ALL ALONG

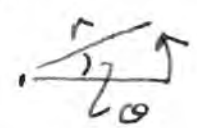
$$U_g = m_1 g y_1 + m_2 g y_2 + \dots = g (m_1 y_1 + m_2 y_2 + \dots) = M_T g y_{\text{cm}}$$

ROTATIONAL KINETIC ENERGY



$$\omega = \frac{2\pi}{T} \leftarrow \text{PERIOD FOR ONE ROTATIONAL}$$

$$\omega r = \frac{2\pi r}{T} = v \leftarrow \text{angular speed}$$



$$\theta = \frac{s}{r} \quad (\theta \text{ in RADIANS})$$

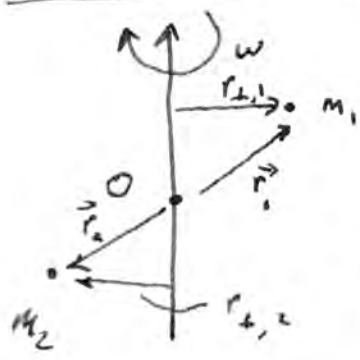
$$\theta r = s$$

$$\frac{d(\theta r)}{dt} = \frac{ds}{dt}$$

CONSTANT r \rightarrow

$$r \frac{d\theta}{dt} = v$$

$$r\omega = v$$



$$v_1 = \omega r_{\perp 1}$$

$$v_2 = \omega r_{\perp 2}$$

KINETIC ENERGY

$$K_{\text{rot}} = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} m_2 v_2^2 + \frac{1}{2} m_3 v_3^2 + \dots$$

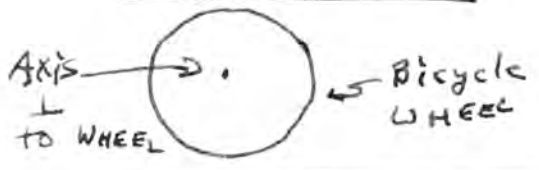
$$K_{\text{rot}} = \frac{1}{2} m_1 r_{\perp 1}^2 \omega^2 + \frac{1}{2} m_2 r_{\perp 2}^2 \omega^2 + \frac{1}{2} m_3 r_{\perp 3}^2 \omega^2 + \dots$$

$$= \frac{1}{2} I \omega^2 \quad I = \text{MOMENT OF INERTIA}$$

$$I = m_1 r_{\perp 1}^2 + m_2 r_{\perp 2}^2 + m_3 r_{\perp 3}^2 + \dots$$

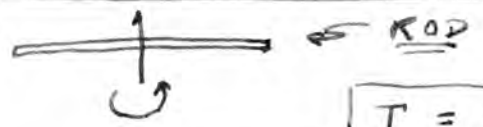
$$K_{\text{rot}} = \frac{1}{2} I \omega^2$$

EXAMPLES

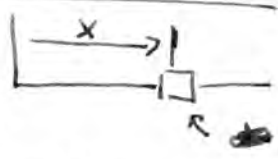


$$I = R^2 (m_1 + m_2 + m_3 + \dots) \quad (\text{HOOP})$$

$$I = MR^2$$



$$\rho = \frac{M}{L} \quad \text{UNIFORM}$$



$$\Delta m = \rho \Delta x$$

$$I = \sum_i \Delta m_i x_i^2$$

$$\Delta m_i = \Delta m = \rho \Delta x \quad (\text{UNIFORM})$$

$$I = \frac{M}{L} \sum_i x_i^2 \Delta x$$

$$I = \frac{M}{L} \left\{ \sum x_i^2 \Delta x \right\} \text{ WHAT IS THIS?}$$

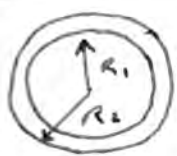
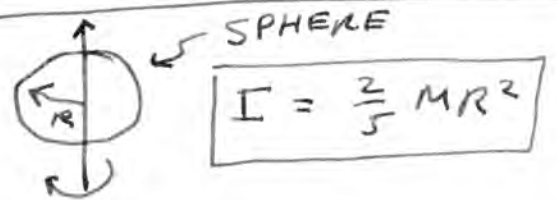
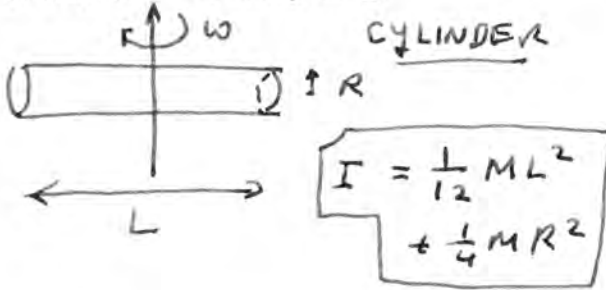
L 20
p.5

$$= \frac{M}{L} \int_{-L/2}^{+L/2} x^2 dx$$

THIN ROD $\Delta x \rightarrow 0$
(INTEGRAL)
 $I = \frac{1}{12} ML^2$

$$= \frac{M}{L} \left[\frac{x^3}{3} \right]_{-L/2}^{+L/2} = \frac{M}{L} \left[\frac{L^3}{24} - \frac{(-L)^3}{24} \right] = \frac{1}{12} ML^2$$

COMMON SHAPES



STRIP AREA = $\pi[R_2^2 - R_1^2]$ Let $R_2 = (R_1 + \Delta R)$
 Let $\Delta R \rightarrow 0$ $\rightarrow R_2^2 = R_1^2 + 2\Delta R R_1 + \Delta R^2$
~~AREA~~ AREA = $\pi 2\Delta R R_1$ $R_2^2 = R_1^2 + 2\Delta R R_1$

$$dA = 2\pi r dr$$

$$\rho = \frac{M}{\pi R^2}$$



$$dm = \rho dA$$

$$I = \sum_i \Delta m_i r_i^2 \Rightarrow \frac{M}{\pi R^2} \int_0^R r^2 2\pi r dr$$

$$I = \frac{MR^2}{2}$$

$$= \frac{2M}{R^2} \left[\frac{r^4}{4} \right]_0^R = \frac{MR^2}{2}$$

OTHERS DONE THE SAME WAY