

# Optical Absorption and Emission

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# Outline

- Applications of solid-state optics: LED, etc.
- Mechanism for light emission and absorption.
- Theory.
- Recent developments in nanoscience:
  - Luminescence in solutions of CdSe nanocrystals.
  - Nanomotors based on azobenzene.

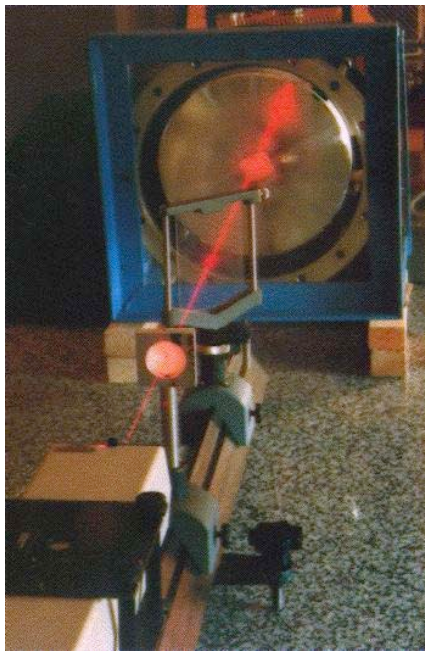
# Why Study Optical Properties?

- One can test different theories for the interaction of light and matter and understand the physical phenomena involved.
- Knowing the physics behind emission/absorption of light, one can design new materials and devices.

## Some Applications of Solid State Optics:



Solar cells



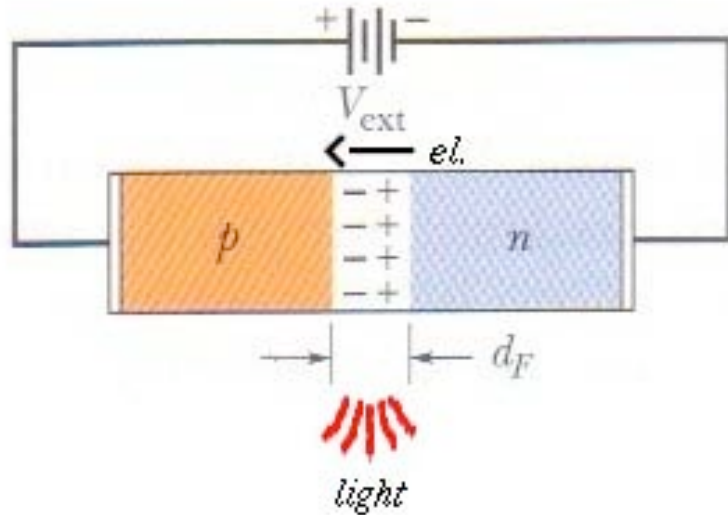
Lasers

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LEDs

# The Physics of a Laser-Emitting Diode (LED)



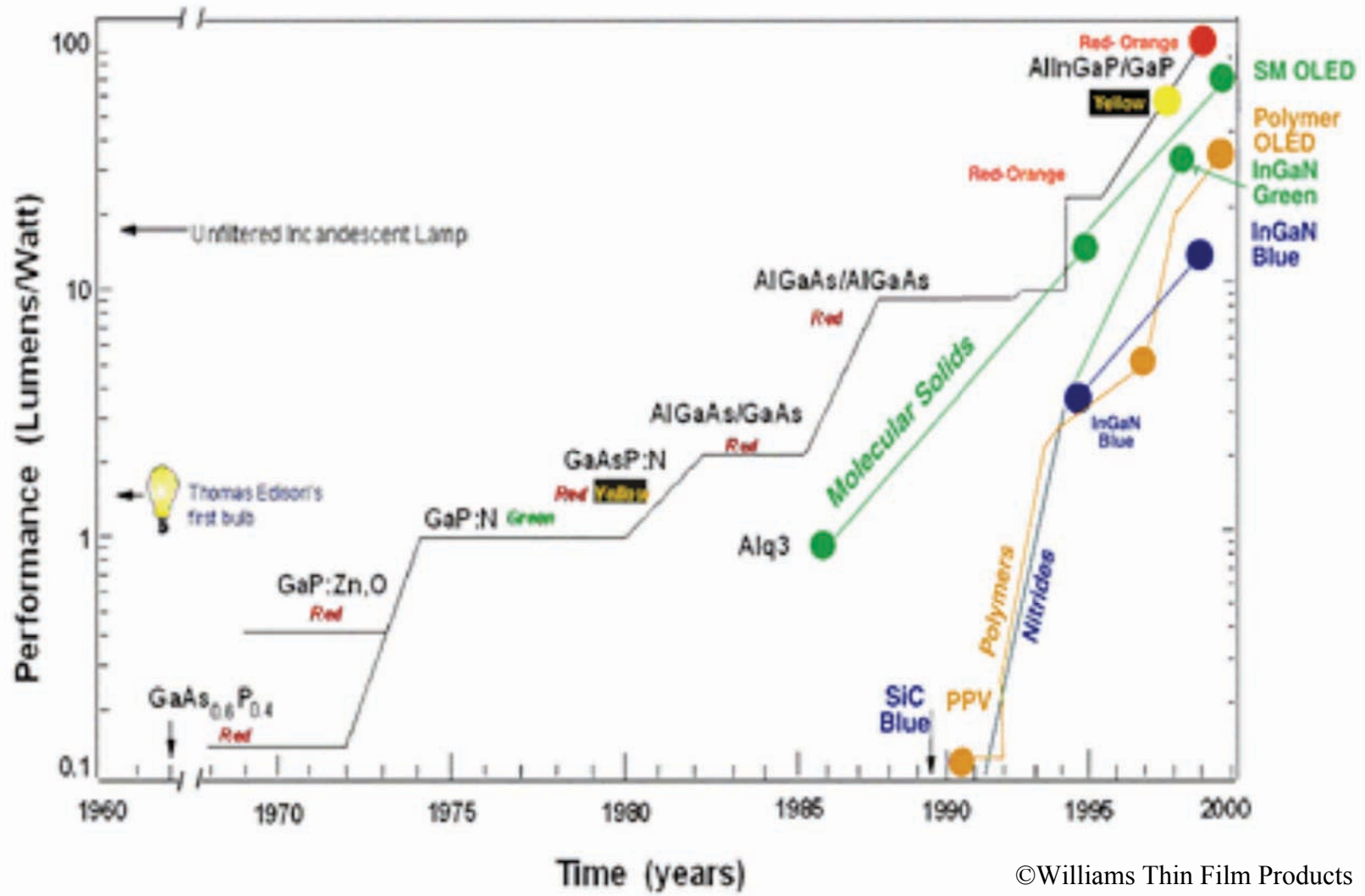
Cheap, low-power source of monochromatic light.

The first LED was a p-n junction made of doped GaAsP.

The semiconductor in the junction defines the color:  
**AlGaAs**, **GaAsP**, **GaN**, **GaP**, **ZnSe**, etc.

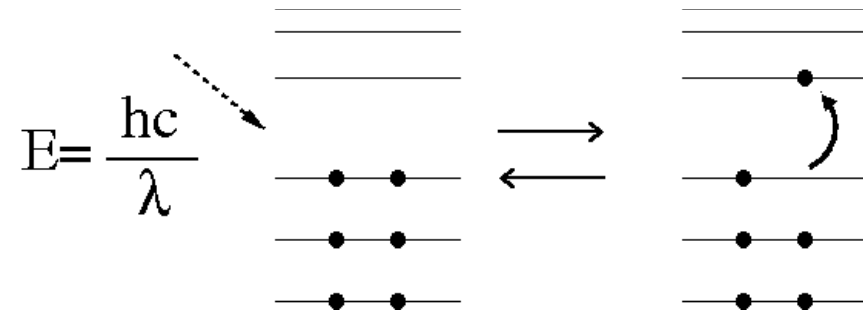
AlGaAs LEDs also emit infrared light

# Evolution of LED Performance



# How Do Materials Absorb and Emit Light?

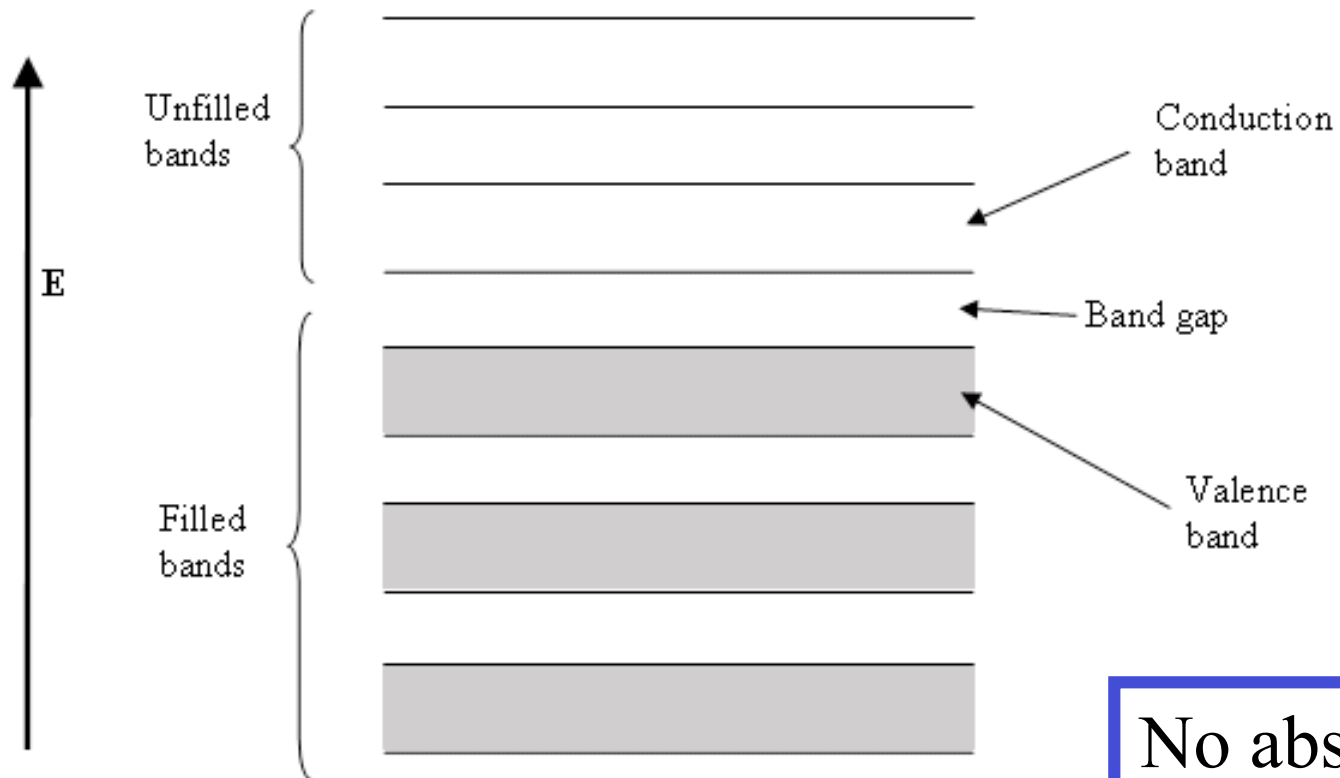
Simple picture: Electrons gain energy from light and undergo a transition from an occupied orbital to an empty orbital.



color	wavelength interval	energy interval
red	~ 625-740 nm	~ 1.7-2.0 eV
orange	~ 590-625 nm	~ 2.0-2.1 eV
yellow	~ 565-590 nm	~ 2.1-2.2 eV
green	~ 500-565 nm	~ 2.2-2.5 eV
cyan	~ 485-500 nm	~ 2.5-2.7 eV
blue	~ 440-485 nm	~ 2.7-2.8 eV
violet	~ 380-440 nm	~ 2.8-3.3 eV

No absorption  
with energy  
below the Gap!

In solids, orbitals split up in bands.



No absorption  
with energy  
below the Gap!



# But not all materials have an energy gap...

Metals have no gap.  
They absorb light at all frequencies. Opaque.



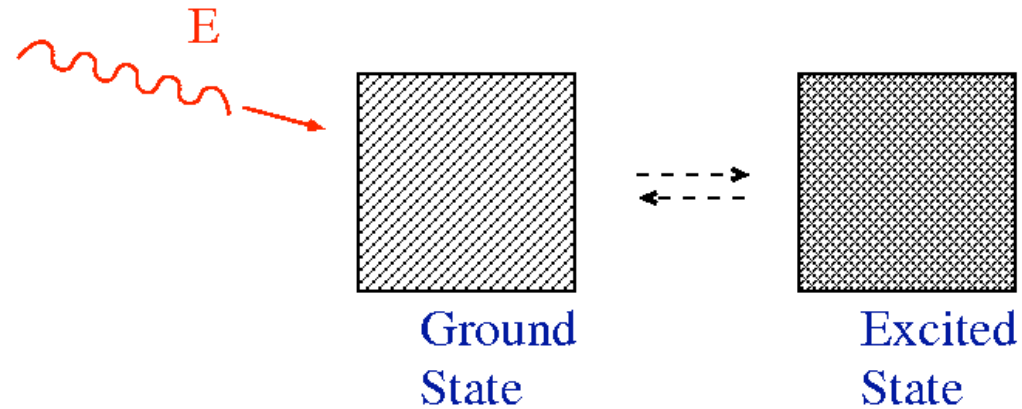
© Alibaba.com Corp.



Insulators have gap:

- water, gap = 6.7 eV
- glass, gap  $> 5$  eV (colored glass is obtained by mixing metals and other impurities).

Failure of the simple picture: It does not include interactions among electrons.



Direct approach: solve the Schrödinger equation

$$H\Psi = i(\partial/\partial t) \Psi$$

Are there simpler methods?

# First-principles Calculations

**Optical absorption can be predicted very accurately from numerical simulations with no input information except for the chemical composition of the material.**

Various theories for electron-electron interactions in the excited state:

- semi-empirical models;
- Time-dependent density-functional theory (TDDFT);
- Bethe-Salpeter equation (BSE);
- Multiconfiguration methods.

# Overview of the Theory:

Density Functional Theory: Electrons in the Ground State



```
graph TD; A[Density Functional Theory: Electrons in the Ground State] --> B[Time-dependent DFT]; A --> C[Bethe-Salpeter Equation]; B --> D[Natural extension to time-dependent processes. Can be rewritten as an eigenvalue problem.]; C --> E[Eigenvalue equation. Eigenvalues determine the normal modes of excitation in the medium.]
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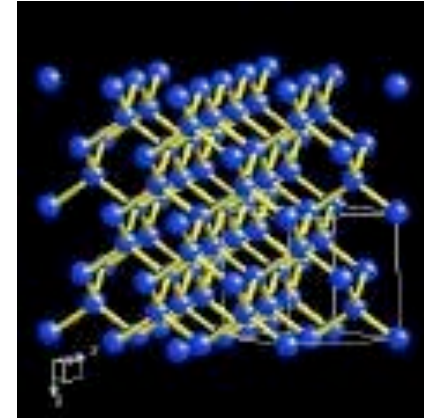
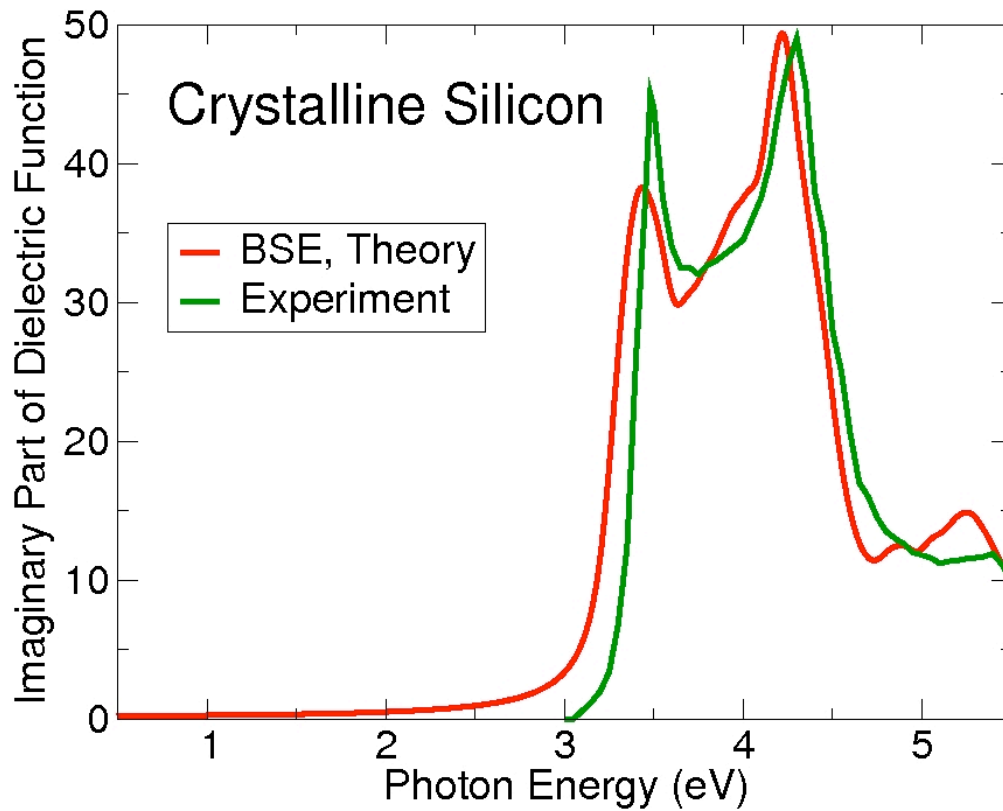
Time-dependent DFT

Bethe-Salpeter Equation

Natural extension to time-dependent processes.  
Can be rewritten as an eigenvalue problem.

Eigenvalue equation.  
Eigenvalues determine the normal modes of excitation in the medium.

# Accurate description of solid materials

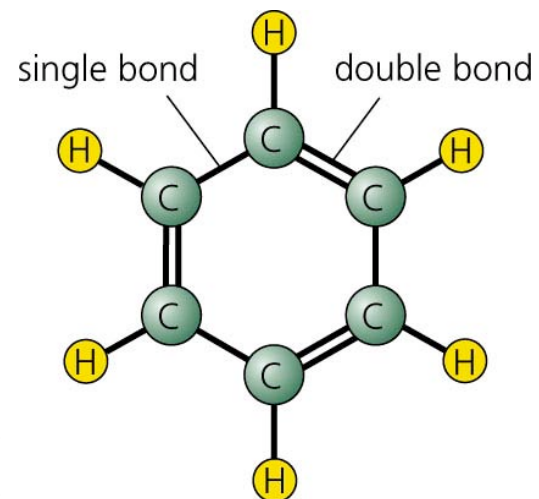
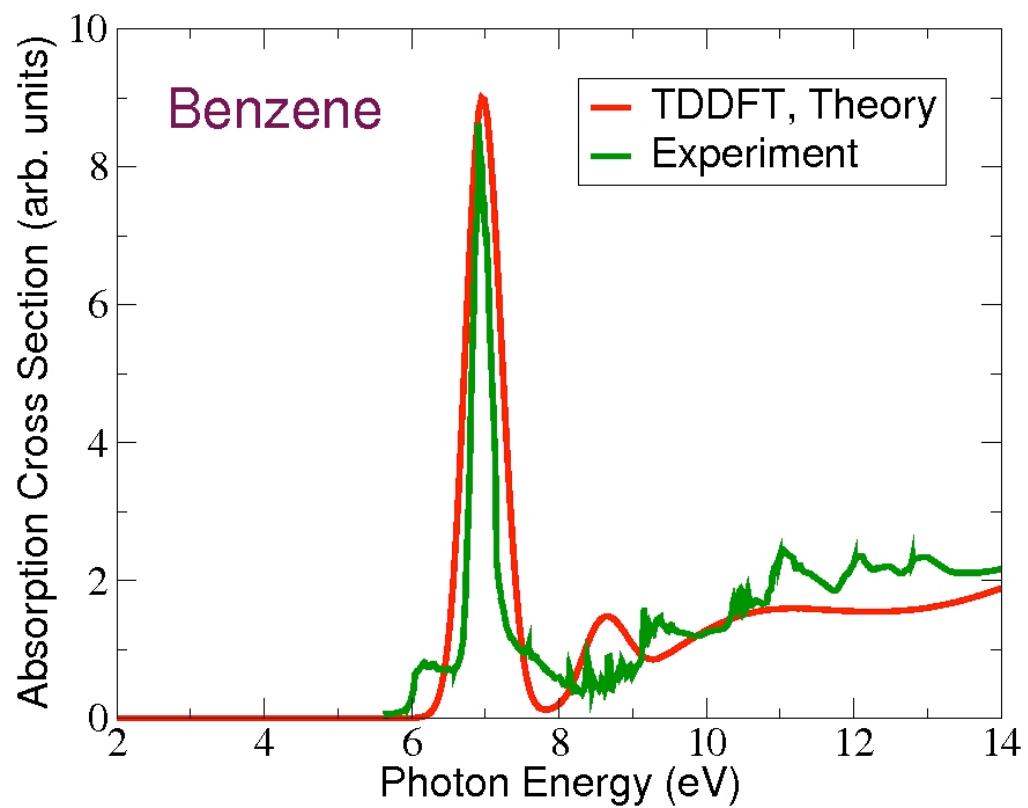


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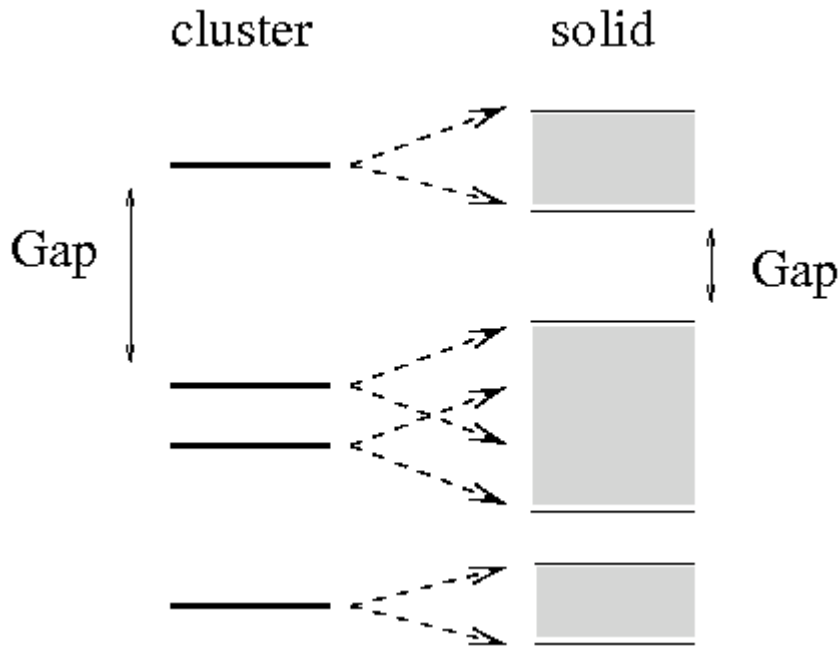
Single Crystal Silicon Ingot

# Accurate description of molecules



# Quantum confinement in nanosystems

Electrons confined to a nanocluster behave differently from free electrons in a solid.



- Gap widening
- Surface effects
- Reduced screening

# CdSe nanocrystals

$\text{Cd}_n\text{Se}_m$ ,  $n+m = 10^3$  to  $10^4$  atoms



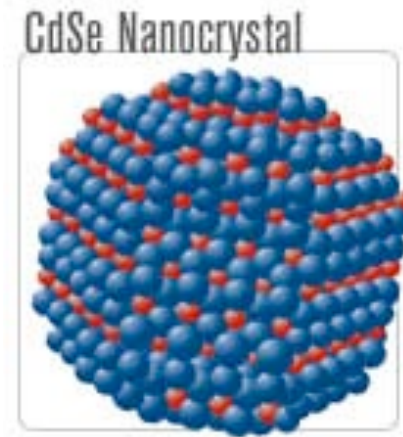
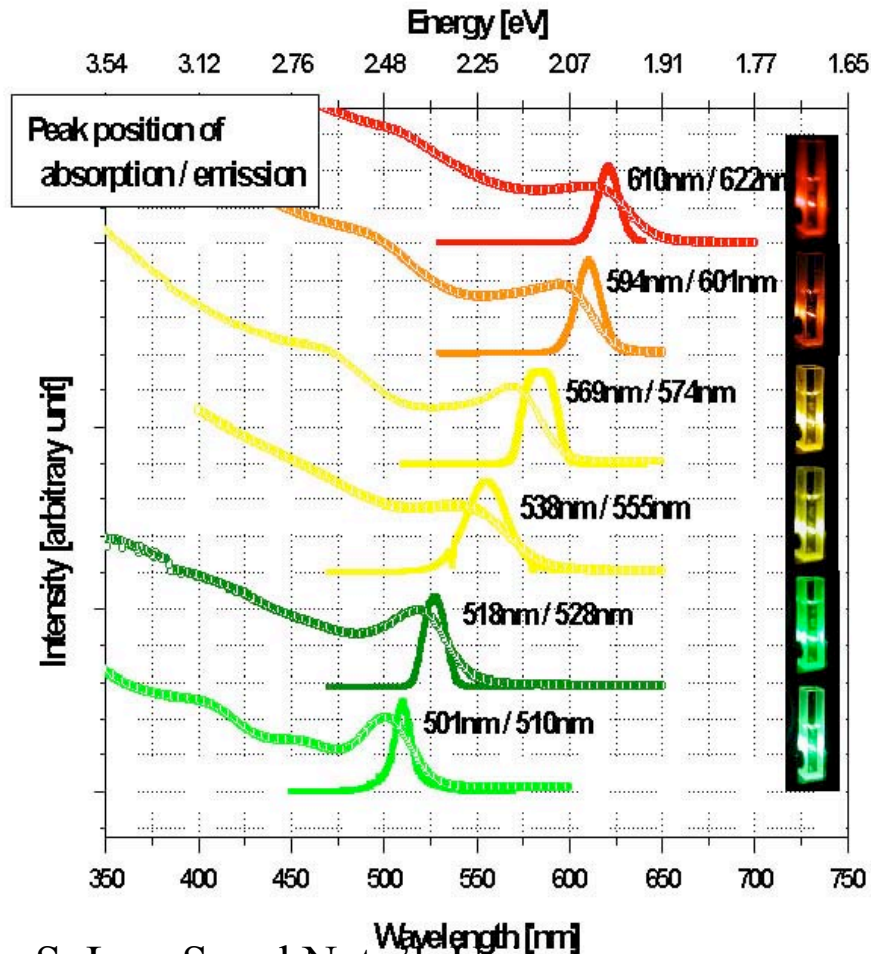
Cluster Size  $\longrightarrow$

Optical activity is highly sensitive to the size of the dot.

Blue CdS/ZnS nanocrystal lasers, Bawendi *et al.* (2005)

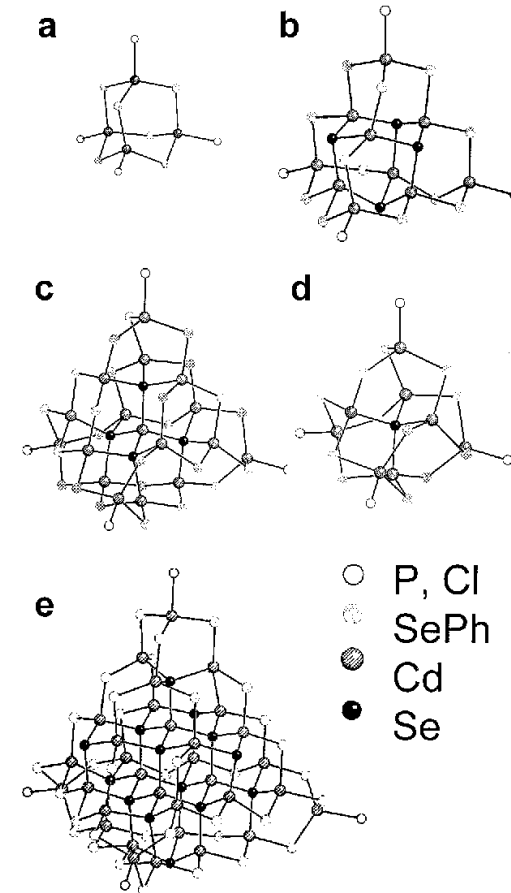
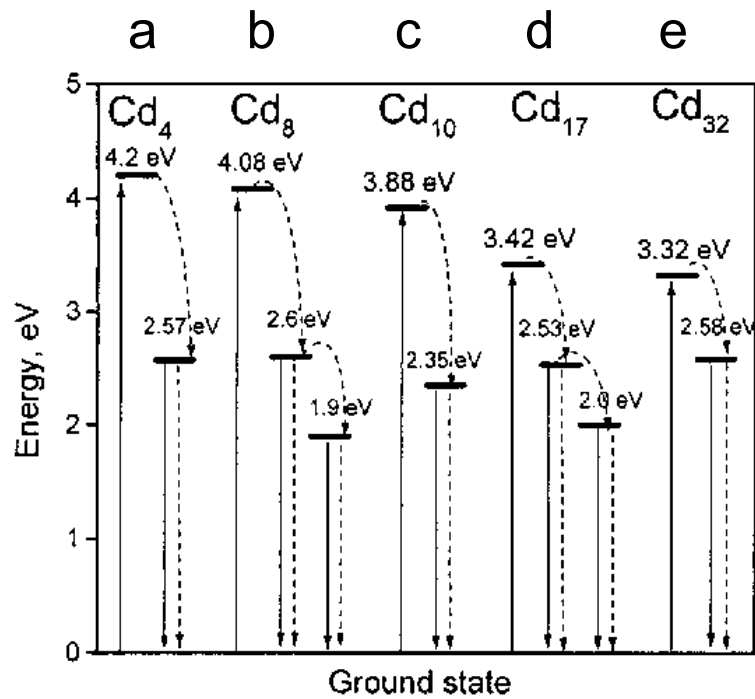


# ✓ Absorption & Emission Spectra of CdSe Quantum Dot



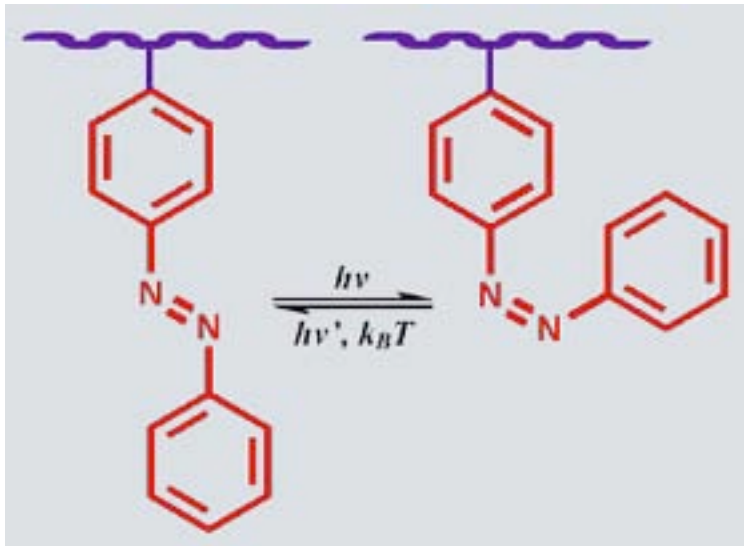
S. Lee, Seoul Natn'l Univ.

Small nanoclusters do not necessary have spherical shape



Banin *et al.* (2001)

# Azobenzene ( $C_{12}H_{10}N_2$ )

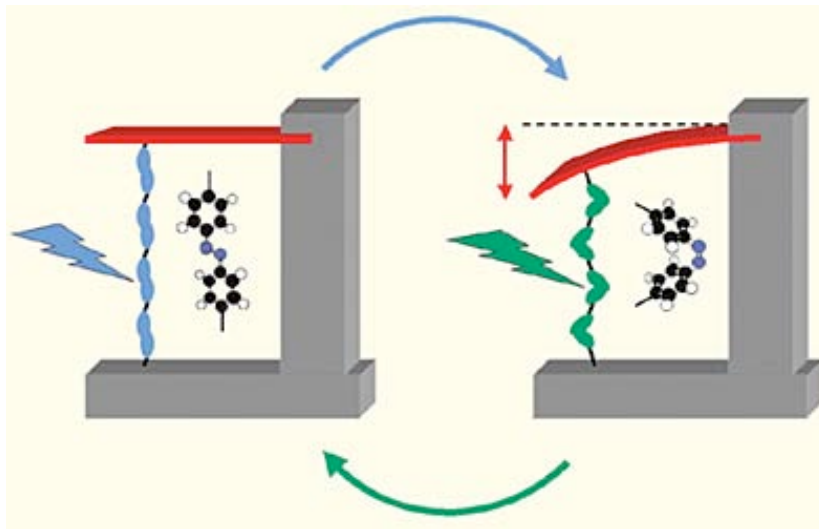


Shows photo- or thermo-induced isomerization.

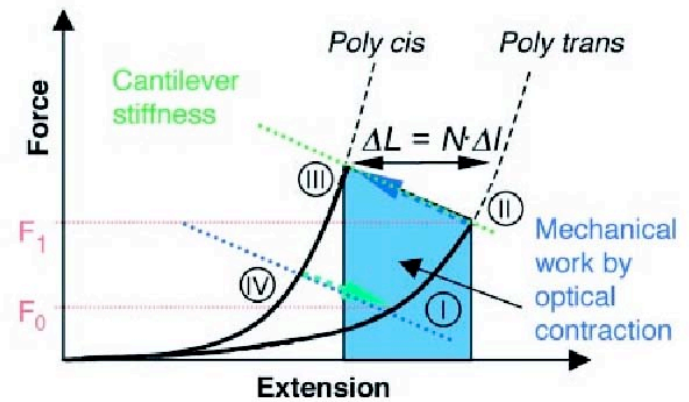
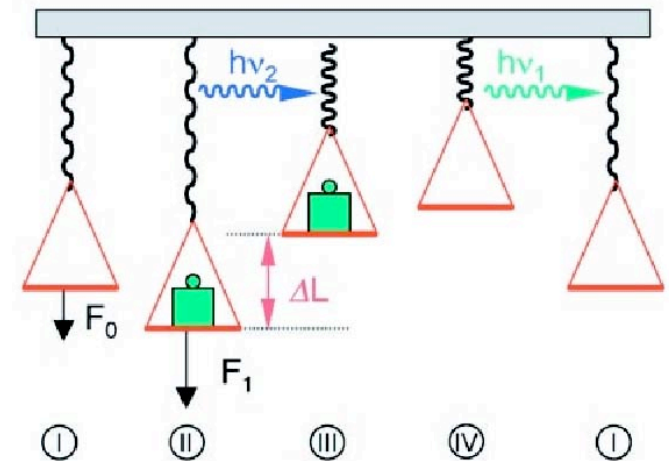
Length changes by 30% during isomerization.

“engine” in a nanomotor.

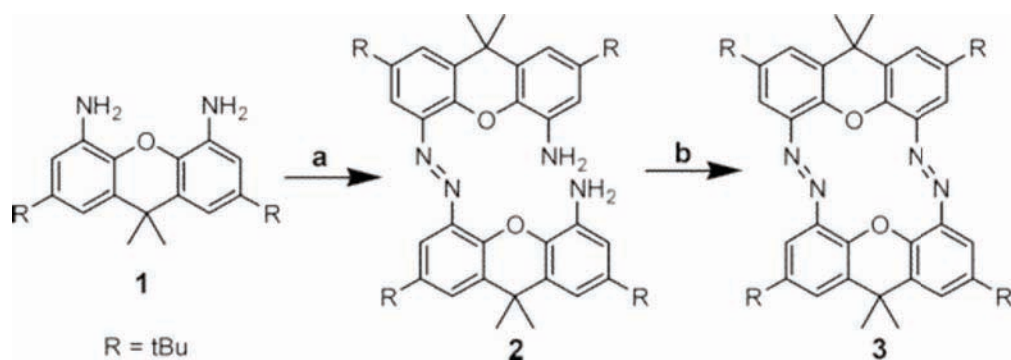
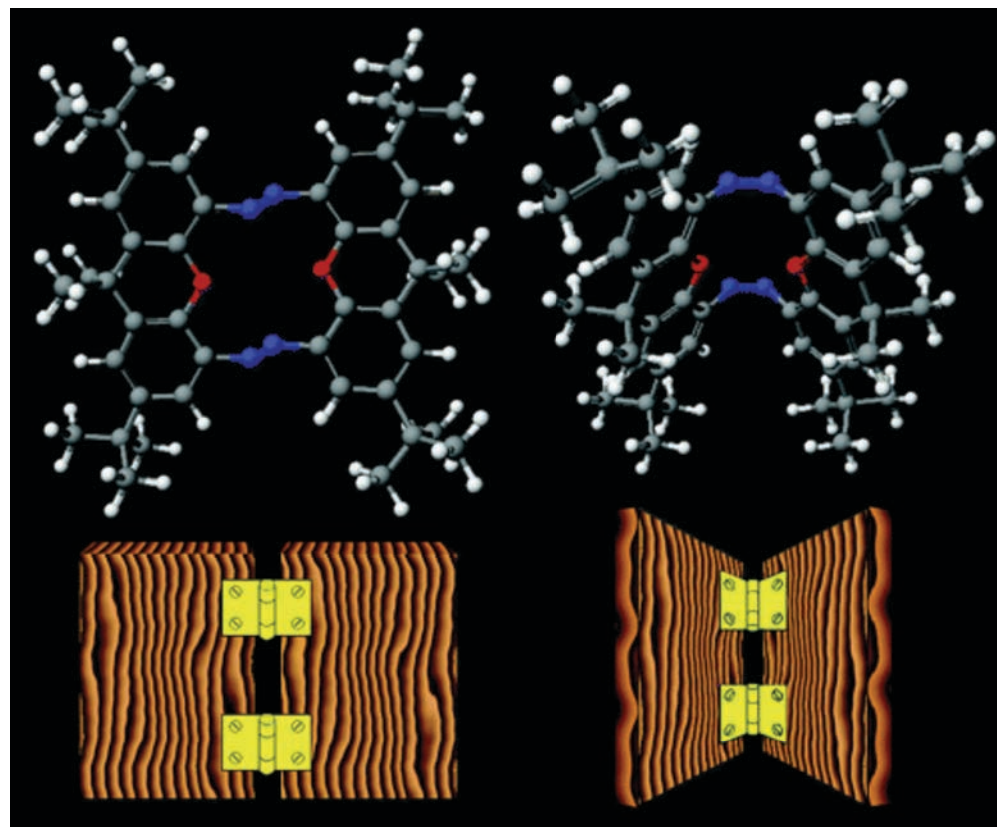
# Nanocantilever from polymerized azobenzene



Gaub *et al.* (2002).

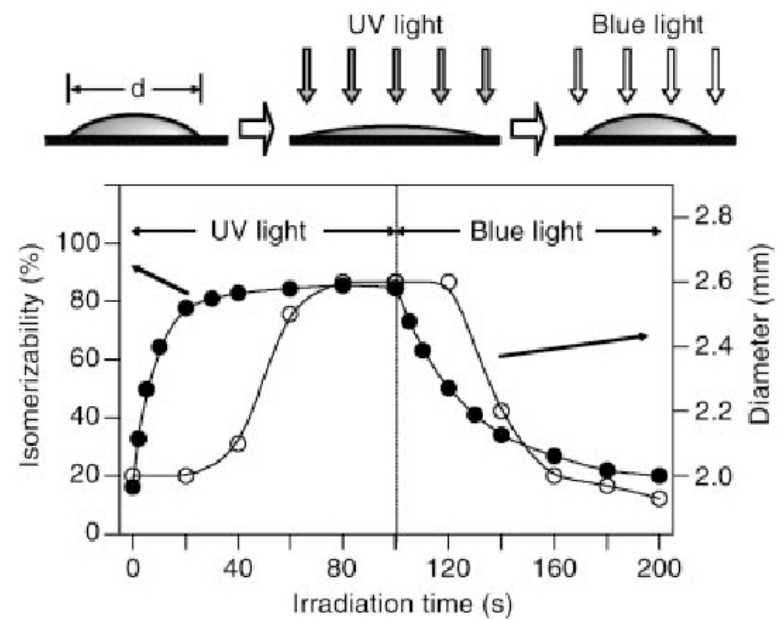
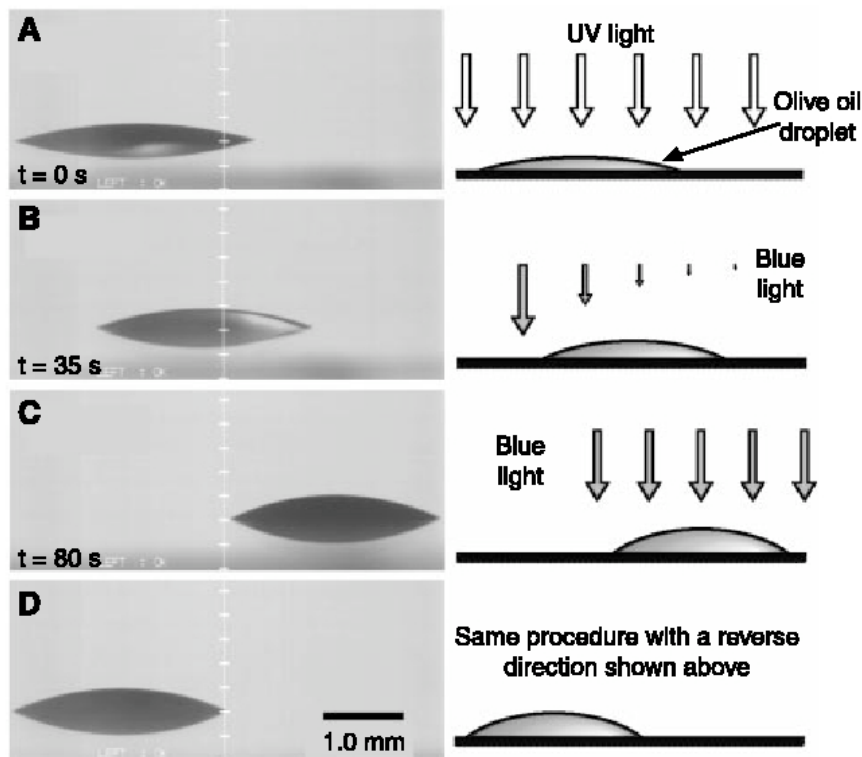


Light-driven  
molecular hinge.



Norikane and Tamaoki  
(2004)

# Light-driven motion of liquids on a photoresponsive surface



Ichimura *et al.* (2000)